

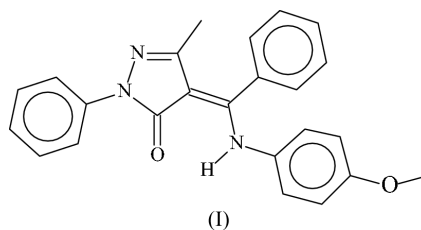
4-[(*Z*)-(4-Methoxyanilino)phenylmethylene]-  
5-methyl-2-phenyl-2*H*-pyrazol-3(4*H*)-oneFeng Bao,<sup>a</sup> Xing-Qiang Lü,<sup>a</sup>  
Yu-Qin Qiao,<sup>a</sup> Qing Wu<sup>a</sup> and  
Seik Weng Ng<sup>b\*</sup><sup>a</sup>School of Chemistry and Chemical Engineering,  
Sun Yat-Sen University, Guangzhou 510275,  
People's Republic of China, and <sup>b</sup>Department of  
Chemistry, University of Malaya, 50603 Kuala  
Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

## Key indicators

Single-crystal X-ray study  
*T* = 298 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$   
*R* factor = 0.040  
*wR* factor = 0.124  
Data-to-parameter ratio = 16.2For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The crystal structure of the title compound,  $\text{C}_{24}\text{H}_{21}\text{N}_3\text{O}_2$ , features a central pyrazole ring; the NH unit interacts with the  $\text{C}=\text{O}$  unit through an intramolecular hydrogen bond [ $\text{N} \cdots \text{O} = 2.714 (1) \text{ \AA}$ ].

## Comment

This study is a continuation of our study of 4-[(*Z*)-(2-aryl-amino)phenylmethylene]-5-methyl-2-phenyl-2*H*-pyrazol-3-ones, which are readily synthesized by condensing 4-benzoyl-3-methyl-1-phenyl-5-pyrazolone with a primary amine (Bao, Lü, Wu, Kang & Ng, 2004; Bao, Lü, Wu, & Ng, 2004; Jiang *et al.*, 2004). A characteristic of such pyrazolones is the short intramolecular hydrogen bond between the amino NH unit and the carbonyl  $\text{C}=\text{O}$  unit; the nature of the organic group connected to the amino group does not appear to have a significant effect on the bond unless the group itself possesses sites that are capable of other interactions. The 4-methoxy-phenyl derivative, (I) (Fig. 1), similarly exists as a monomeric molecule that features an intramolecular hydrogen bond [2.714 (1)  $\text{ \AA}$ ].

## Experimental

4-Benzoyl-3-methyl-1-phenyl-5-pyrazolone (2.20 g, 7.9 mmol) and *p*-methoxyaniline (0.99 g, 8.0 mmol) were dissolved in formic acid (35 ml). The solution was heated under reflux for 8 h. The solvent was removed and the pure product obtained upon recrystallization from a 1:1 ethanol/*n*-heptane mixture (45 ml) in about 80% yield. Crystals were grown from ethanol. Analysis calculated for  $\text{C}_{24}\text{H}_{21}\text{N}_3\text{O}_2$ : C 75.20, H 5.48, N 10.97%; found: C 75.30, H 5.39, N 10.68%.

## Crystal data

 $\text{C}_{24}\text{H}_{21}\text{N}_3\text{O}_2$   
*M<sub>r</sub>* = 383.44  
Triclinic, *P1*  
*a* = 6.8262 (9)  $\text{ \AA}$   
*b* = 10.951 (1)  $\text{ \AA}$   
*c* = 14.761 (1)  $\text{ \AA}$   
 $\alpha$  = 97.259 (2)°  
 $\beta$  = 103.386 (2)°  
 $\gamma$  = 106.018 (2)°  
*V* = 1010.2 (2)  $\text{ \AA}^3$ *Z* = 2  
*D<sub>x</sub>* = 1.261  $\text{ Mg m}^{-3}$   
Mo *K*α radiation  
Cell parameters from 904  
reflections  
 $\theta$  = 2.0–27.1°  
 $\mu$  = 0.08  $\text{ mm}^{-1}$   
*T* = 298 (2) K  
Block, yellow  
0.50 × 0.50 × 0.20 mm

Data collection

Bruker SMART area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: none  
 8643 measured reflections  
 4350 independent reflections

3303 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.013$   
 $\theta_{\text{max}} = 27.1^\circ$   
 $h = -8 \rightarrow 8$   
 $k = -13 \rightarrow 13$   
 $l = -18 \rightarrow 18$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.040$   
 $wR(F^2) = 0.124$   
 $S = 1.01$   
 4350 reflections  
 268 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0692P)^2 + 0.1202P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$

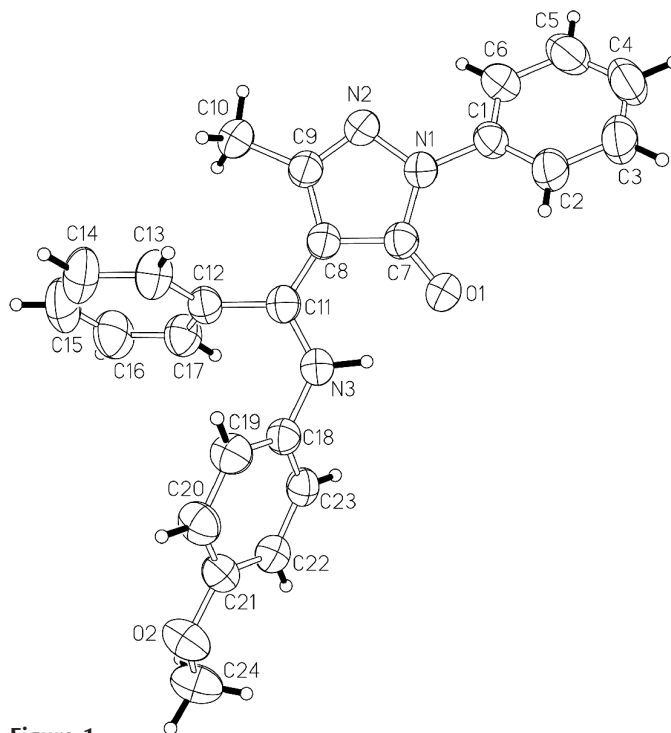


Figure 1 ORTEPII (Johnson, 1976) plot of (I), with ellipsoids drawn at the 50% probability level. H atoms are drawn as spheres of arbitrary radii.

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

|          |           |           |           |
|----------|-----------|-----------|-----------|
| O1—C7    | 1.244 (2) | C7—C8     | 1.444 (2) |
| N1—N2    | 1.397 (2) | C8—C11    | 1.391 (2) |
| N1—C7    | 1.376 (2) | C8—C9     | 1.433 (2) |
| N1—C1    | 1.414 (2) | C9—C10    | 1.490 (2) |
| N2—C9    | 1.303 (2) |           |           |
| N2—N1—C1 | 118.7 (1) | C7—C8—C9  | 105.2 (1) |
| N2—N1—C7 | 112.2 (1) | C7—C8—C11 | 122.2 (1) |
| C1—N1—C7 | 128.5 (1) | C9—C8—C11 | 132.2 (1) |
| C9—N2—N1 | 106.4 (1) | N2—C9—C8  | 111.7 (1) |
| O1—C7—N1 | 126.2 (1) | N2—C9—C10 | 118.5 (1) |
| N1—C7—C8 | 104.4 (1) | C8—C9—C10 | 129.8 (1) |
| O1—C7—C8 | 129.3 (1) |           |           |

Table 2

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

| $D-H\cdots A$      | $D-H$    | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------|----------|-------------|-------------|---------------|
| N3—H3n $\cdots$ O1 | 0.87 (1) | 1.96 (1)    | 2.714 (1)   | 145 (2)       |

The H atoms were placed at calculated positions [ $C-H = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for the aromatic H atoms and  $C-H = 0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for the methyl H atoms] and included in the refinement in the riding-model approximation. The amino H atom was located and refined with an  $N-H$  distance restraint of  $0.85 (1) \text{ \AA}$ .

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine

structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

The authors thank the National Natural Science Foundation of China, the Natural Science Foundation of Guangdong Province, Sun Yat-Sen University and the University of Malaya for supporting this work.

References

Bao, F., Lü, X.-Q., Wu, Q., Kang, B.-S. & Ng, S. W. (2004). *Acta Cryst.* **E60**, o155–o156.  
 Bao, F., Lü, X.-Q., Wu, Q. & Ng, S. W. (2004). *Acta Cryst.* **E60**, o1244–o1245.  
 Bruker (2001). SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Jiang, J.-J., Lü, X.-Q., Bao, F., Kang, B.-S. & Ng, S. W. (2004). *Acta Cryst.* **E60**, o1149–o1150.  
 Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.  
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.